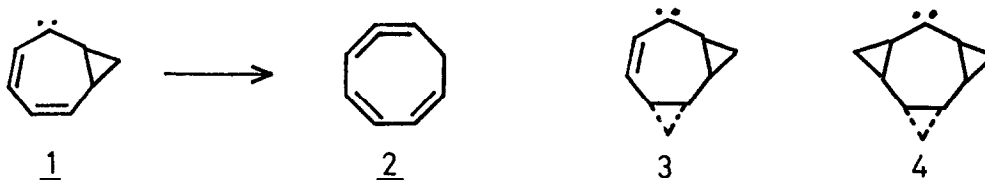


GENERATION AND REACTIONS OF ANTI-TRICYCLO[6.1.0.0<sup>2,4</sup>]NON-5-EN-7-YLIDENE AND ANTI, ANTI-TETRACYCLO[7.1.0.0<sup>2,4</sup>.0<sup>5,7</sup>]-DECAN-8-YLIDENE

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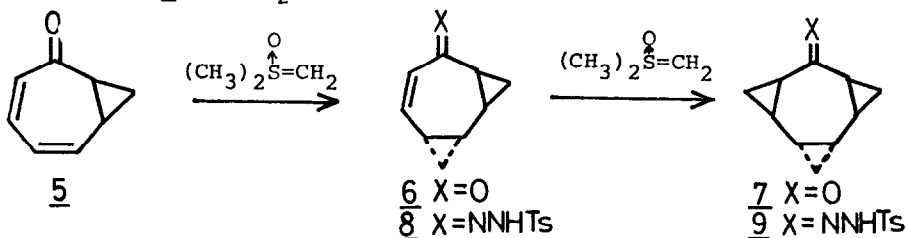
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Due to high reactivity of carbene and to strain of cyclopropane ring, cyclopropyl carbenes undergo fragmentations and rearrangements rather than additions and insertions.<sup>1</sup> We have recently reported the facile rearrangement of bicyclo-[5.1.0]octa-2,4-diene-6-ylidene 1 to cycloocta-1,2,4,6-tetraene 2.<sup>2</sup> We here describe the generation of the title carbenes, 3 and 4, whose behaviors differ each other and also from those of 1.



Preparations of the precursors:

Reaction of 2,3- $\sigma$ -homotropone 5<sup>3</sup> with one equiv. of dimethylsulfoxonium methylide in dimethyl sulfoxide gave anti-2,3;4,5-bis- $\sigma$ -homotropone 6 in 72-86% yield (b.p. 90-92°C/0.3 torr.). Further treatment of 6 with one equiv. of the reagent or treatment of 5 with two equiv. of the reagent gave anti, anti-2,3;4,5;6,7-tris- $\sigma$ -homotropone 7 (m.p. 65°C) in 30-45% yield. The assigned structures were deduced from the spectral data (Table 1) and some chemical bases.<sup>4</sup> The homotropone 6 was then converted into the tosyl hydrazone 8 (m.p. 135°C) in 51% yield by reaction with *p*-toluenesulfonyl hydrazide in refluxing benzene (NH<sub>4</sub>OAc as a catalyst). The tosyl hydrazone 9 (m.p. 179°C decomp.) was obtained from 7 in 98% yield (*p*-TsNHNH<sub>2</sub>, pyridine, 60°C).



anti-Tricyclo[6.1.0.0<sup>2,4</sup>]non-5-en-7-ylidene 3 (Scheme 1):

Irradiation of the sodium salt of 8, obtained as solids by treatment with sodium hydride, in tetrahydrofuran (THF) in the presence of either dimethyl fumarate (DMF) or maleate (DMM) (five equiv.) yielded two trans adducts, 10a (m.p. 88°C) and 10b (m.p. 118°C) (27%:30% (DMF); 13%:15% (DMM)). The adducts were also obtained in similar yields and ratios, respectively, on a thermal condition of refluxing dioxane. The trans stereochemistry was suggested by the coupling constants between the vicinal cyclopropane protons (6.3 Hz in 10a and 6.0 Hz in 10b; see Table 1).<sup>5</sup> Cycloheptatrienylidene has been reported to give only trans adduct with both the olefins, though it adds to fumaro- and malei-nitrile stereospecifically at careful conditions.<sup>6,7</sup> It is yet uncertain which adduct is assignable to the structure 10a or 10b.

On the other hand, when thermally generated in the presence of diphenyliso-benzofuran (DPIBF), the carbene 3 gave two, stereoisomeric Diels-Alder adducts, 12a (18%, m.p. 210°C) and 12b (33%, m.p. 127°C), via transformation to highly strained dehydro-tris- $\sigma$ -homobenzenes (anti-tetracyclo[6.1.0.0<sup>2,4</sup>.0<sup>5,7</sup>]non-9-enes), 11a and 11b. The PMR spectra indicate the absence of olefin protons. The PMR signals at  $\delta$  2.67 in 12a and 2.47 in 12b could be assigned to H<sub>b</sub> in each adduct (see the Figure), whose lower field appearance relative to the other cyclopropane protons suggests its proximity to the oxygen bridge. Such a proximity effect has been observed in related systems.<sup>8</sup> The relatively small  $J_{\text{H}_a\text{-H}_b}$  in both the adducts (3.5 Hz in 12a and 3.2 Hz in 12b) indicate trans relationship of these protons.<sup>5</sup> The large difference in the  $J_{\text{H}_b\text{-H}_c}$  (8.0 Hz in 12a and nearly zero in 12b) leads to a conclusion that the relationships between these protons of 12a and 12b are cis and trans, respectively, since the cyclohexane rings can be regarded essentially planar.<sup>9</sup>

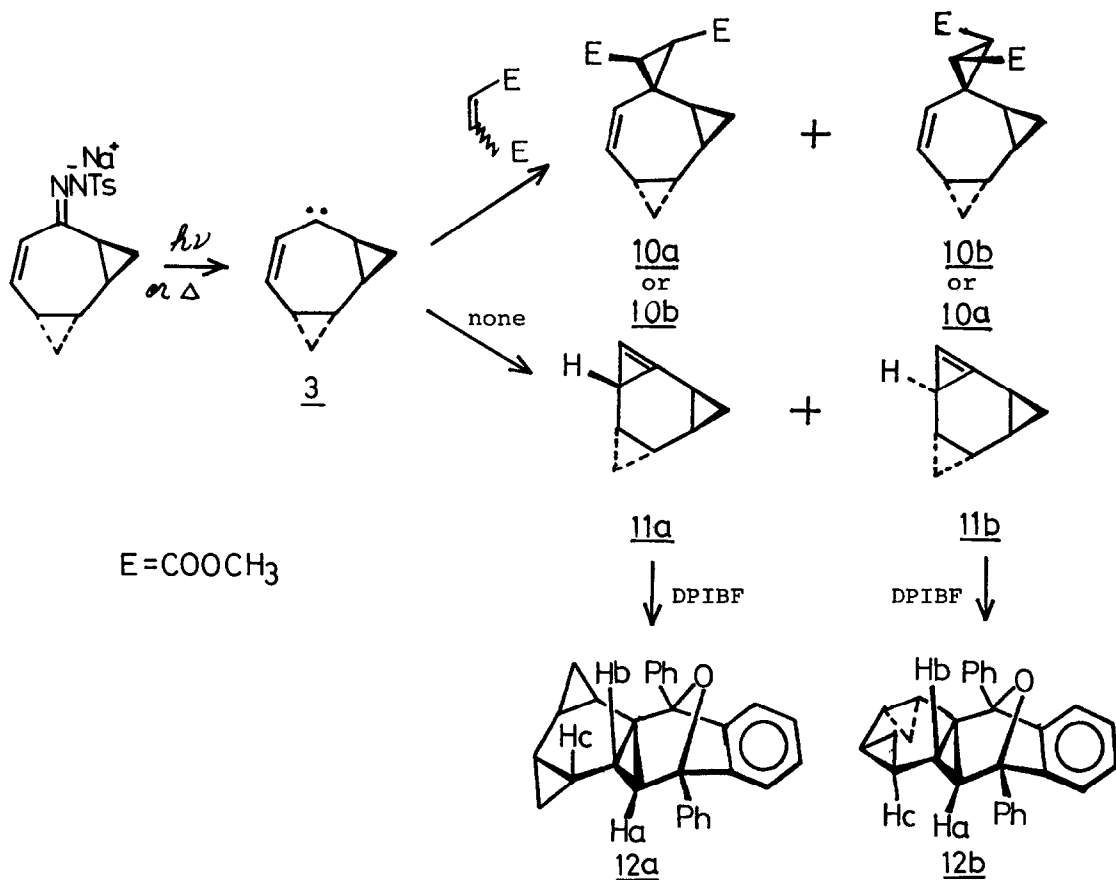
In the absence of trapping agents, the carbene 3 gave only a trace amount of volatile compounds whose structures have been yet uncertain.

Intramolecular cycloadditions of carbenes to double bonds leading to cyclopropenes are well documented.<sup>1</sup> As a related system, fused cyclopropenes have been strongly supported as the intermediates in carbene-carbene rearrangement of cycloheptatrienylidenes and benzyl carbenes.<sup>8, 10</sup>

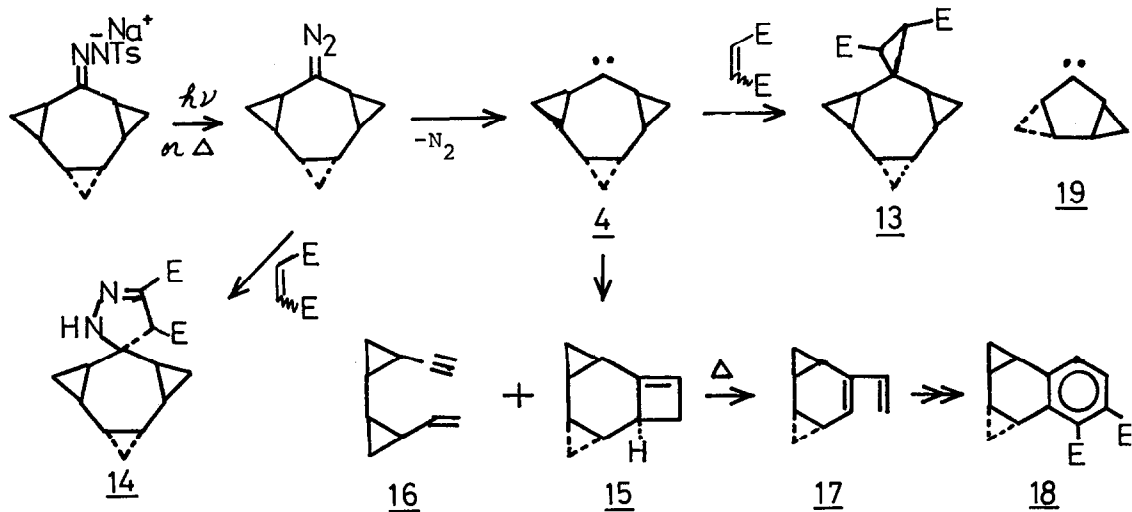
anti, anti-Tetracyclo[7.1.0.0<sup>2,4</sup>.0<sup>5,7</sup>]decan-8-ylidene 4 (Scheme 2):

Thermal decomposition of the sodium salt of 9 in refluxing dioxane in the presence of either DMF or DMM gave the adduct 13 (1.2-3.5%, m.p. 74°C) and the dihydropyrazole 14 (90.8-91.4%, m.p. 156°C). The stereochemistries were tentatively assigned as shown in view of the results on 3 and steric hindrance.

However, irradiation of the salt in THF alone yielded two hydrocarbons, 15 (28%, oil) and 16 (3%, oil). The spectral data (Table 1) support their structures. Conclusive evidence on 15 was obtained by its thermal reaction. Heating of 15 in toluene (1.5 h) gave quantitatively the ring opened compound 17 which was in turn led to the bis- $\sigma$ -homonaphthalene 18 (80%) by reaction with dimethyl



Scheme 1



Scheme 2

acetylenedicarboxylate followed by dehydrogenation with DDQ.

The predominant ring contraction of 4 over fragmentation contrasts with the exclusive fragmentation of anti-tricyclo[4.1.0.0<sup>2,4</sup>]heptan-5-ylidene 19.<sup>11</sup> The latter's result may be because the similar ring contraction would produce a highly strained molecule and hence be energetically disfavored.

Table 1. Selected spectral data of the products.

6:  $\nu$  (oil) 1640, 1620  $\text{cm}^{-1}$ ;  $\lambda$  (EtOH) 262.5 nm ( $\log \epsilon$  3.63);  $\delta$  ( $\text{CCl}_4$ ) 0.69 (1H, m), 1.15-2.15 (7H, m), 5.28 (1H, dd,  $J=12.5$ , 1.5 Hz), 6.45<sup>4</sup> (1H, dd, 12.5, 7.3).

7:  $\nu$  (KBr) 1690  $\text{cm}^{-1}$ ;  $\lambda$  (EtOH) 280 nm ( $\log \epsilon$  1.83);  $\delta$  ( $\text{CDCl}_3$ ) 0.03 (1H, m), 0.5-1.2 (9H, m), 2.22 (2H, ddd,  $J=8.5$ , 8.5, 7.5 Hz).

10a:  $\delta$  ( $\text{CDCl}_3$ , 100 MHz) 0.4-1.7 (8H, m), 2.56 (2H, narrow AB quartet,  $J=6.3$  Hz) 3.68 (3H, s), 3.70 (3H, s), 4.80 (1H, d,  $J=12.0$ ), 5.90 (1H, dd, 12.0, 6.5)

10b:  $\delta$  ( $\text{CDCl}_3$ , 100 MHz) 0.3-1.3 (8H, m), 2.52 (2H, narrow AB quartet,  $J=6.0$  Hz), 3.67 (3H, s), 3.73 (3H, s), 5.43 (1H, d, 10.5), 5.80 (1H, br. d, 10.5).

12a:  $\delta$  ( $\text{CDCl}_3$ ) 0-1.30 (8H, m), 1.33 (1H, d,  $J=3.5$  Hz), 2.67 (1H, dd, 8.5, 3.5), 7.1-8.2<sup>3</sup> (14H, m).

12b:  $\delta$  ( $\text{CDCl}_3$ ) -0.42 (1H, ddd,  $J=4.8$ , 4.8, 4.5 Hz), 0-1.10 (7H, m), 1.18 (1H, d, 3.2), 2.47 (1H, d, 3.2), 7.0-7.9 (14H, m).

13:  $\delta$  ( $\text{CDCl}_3$ ) 0-0.5 (12H, m), 3.15 (2H, s), 3.67 (3H, s), 3.83 (3H, s).

14:  $\nu$  (KBr) 3355, 1720, 1700  $\text{cm}^{-1}$ ;  $\delta$  ( $\text{CDCl}_3$ ) 0.2-1.5 (12H, m), 3.73 (3H, s), 3.83 (3H, s), 3.97 (1H, s), 6.25 (1H, br., NH).

15:  $\nu$  (oil) 1640  $\text{cm}^{-1}$ ;  $\delta$  ( $\text{CCl}_4$ ) 0.10 (1H, m), 0.8-1.2 (5H, m), 1.41 (1H, m) m 2.1-2.3 (2H, m), 2.75 (1H, ddd,  $J=12.4$ , 3.8, 3.8 Hz), 5.83 (1H, structured s).

16:  $\nu$  (oil) 3310, 2125, 1635  $\text{cm}^{-1}$ ;  $\delta$  ( $\text{CCl}_4$ ) 0.3-1.5 (8H, m), 1.63 (1H, d,  $J=1.8$  Hz, CH), 4.7-5.6 (3H, m, vinyl).

17:  $\nu$  (oil) 1635, 1600  $\text{cm}^{-1}$ ;  $\lambda$  (EtOH) 254 nm ( $\log \epsilon$  4.18);  $\delta$  ( $\text{CCl}_4$ ) 0.3-1.7 (8H, m), 4.82 (1H, ddd,  $J=10.5$ , 1.6, 0.8 Hz), 5.15 (1H, ddd, 17.5, 1.6, 0.5), 6.55 (1H, br. d, 4.0), 6.20 (1H, dd, 17.5, 10.5)

18:  $\delta$  ( $\text{CCl}_4$ ) 0.5-1.3 (4H, m), 1.6-1.9 (4H, m), 3.80 (6H, s), 7.12 (1H, d,  $J=8.0$  Hz), 7.56 (1H, d, 8.0).

#### REFERENCES AND NOTES

- 1) For a review, see W. Kirmse, "Carbene chemistry", Second Ed., Academic Press, New York, N. Y., 1971.
- 2) M. Oda, Y. Ito, and Y. Kitahara, *Tetrahedron Lett.*, 2587 (1975).
- 3) a) J. D. Holmes and R. Pettit, *J. Amer. Chem. Soc.*, 85, 2531 (1963); b) M. Oda, T. Sato, and Y. Kitahara, *Synthesis*, 721 (1974).
- 4) The details will be reported elsewhere.
- 5) K. L. Williamson, C. A. Lanford, and C. R. Nicholson, *J. Amer. Chem. Soc.*, 86, 762 (1964).
- 6) T. Mukai, T. Nakazawa, and K. Isobe, *Tetrahedron Lett.*, 565 (1968).
- 7) W. M. Jones, B. N. Hamon, R. C. Joines, and C. L. Ennis, *Tetrahedron Lett.*, 3909 (1969).
- 8) T. T. Coburn and W. M. Jones, *J. Amer. Chem. Soc.*, 96, 5218 (1974).
- 9) The cyclohexane ring of tris-homobenzenes have been revealed to be planar by X-ray analyses; see C. Kabuto, M. Yagihara, T. Asao, and Y. Kitahara, *Angew. Chem.*, 85, 860 (1973); and W. Littke and U. Druck, *ibid*, 86, 557 (1974). The dihedral angles estimated in such a model are 0° (cis) and 90° (trans).
- 10) J. P. Mykytka and W. M. Jones, *J. Amer. Chem. Soc.*, 97, 5933 (1975); and references therein.
- 11) W. R. Dobbier, Jr., O. T. Garza, and B. H. Al-Sader, *J. Amer. Chem. Soc.*, 97, 5038 (1975).